DEPT. OF EARTH & ENVIRONMENTAL SCIENCES, UNIVERSITY OF ROCHESTER

Introduction to Atmospheric Chemistry EES 218



Pollution plume at sunset over Western New York viewed from Space Shuttle Discovery, Oct. 21, 2000.

Spring 2023 Syllabus

Last Updated: January 24, 2023

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1 OVERVIEW

Course Location: Harkness Hall 210 **Course Time:** Tu/Th 11:05 AM-12:20 PM

Recitation Location: Hylan Hall 102 **Recitation Time:** W 11:50 AM-1:05 PM

Instructor: Prof. Lee T. Murray E-mail: lee.murray@rochester.edu Office Location: Hutchison Hall 479 Office Hours: Tu 12:30-1:30 PM, or by appointment

Teaching Assistant: Matthew Loman E-mail: mloman@ur.rochester.edu Office Hours: Th 12:30-1:30 PM, or by appointment Office Location: Hutchison Hall 130B

1.1 DESCRIPTION

The atmosphere helps to maintain habitable temperatures on our planet's surface, shields life from destructive cosmic and ultraviolet radiation and contains gases such as oxygen and carbon dioxide, which are essential for life. In this course we will work toward an understanding of several important questions. What is in the Earth's atmosphere? What are the sources and sinks of the most important gases in the atmosphere? How does the atmosphere affect the Earth's surface climate? What is the role of photochemistry in atmospheric composition? How does the atmosphere interact with the land and oceans? How has human activity affected the atmosphere?

1.2 PRE-REQUISITES

Required, unless granted permission by instructor:

- EES 101 (Intro. to Geology), EES 103 (Intro. to Environmental Sci.), or EES 105 (Intro. to Climate Change) or equivalent
- CHM 131 (Intro. College Chemistry I) or equivalent
- MTH 141-142 (Calculus I and II) or equivalent

Recommended, but not required:

- EES 236 (Intro. Atmospheric Physics) or equivalent
- CHM 132 (Intro. College Chemistry II) or equivalent

Permission of instructor is required for majors other than GEO, EVS and ESP.

1.3 MAIN LEARNING GOALS

By the end of the course, students will understand:

- The composition of the modern atmosphere
- The basics of atmospheric structure and circulation
- The concepts of sink, source, and lifetime, and be able to make and use simple box models to describe changes in atmospheric composition
- The atmospheric budgets of the most important trace gases
- The dominant processes and chemical reactions in the stratosphere
- The dominant processes and chemical reactions in the troposphere
- The stratospheric ozone loss problem
- Major surface air-quality issues and their associated control strategies
- Earth's radiative energy budget and how the greenhouse effect works
- The sources and sinks of aerosol particles and how they affect climate
- The coupling between air quality and climate change

2 Schedule

Below is the tentative class schedule, with topics for each class, associated readings, and problem set due dates. Schedule subject to change pending course progress.

Required reading assignments in **bold**. Supplemental or alternative readings in *italics*.

There is no password for the Jacob (1999) text. The password for the Jacob (2023) text is "atmchem".

TUESDAY	THURSDAY	
Jan 10th	12th	1
	Introduction and Overview	
	Key Topics	
	1. History and evolution of Earth's	
	atmospheric composition	
	Reading	
	Syllabus	
	• <i>Jacob</i> (2023) §5.2	
	• Zahnle et al. (2010)	
	• <i>Jacobson</i> (2012) §2	
	 Seinfeld and Pandis (2016) §1.1 	
	Note: Held Tu Jan 17 @ 11:05 in Harkness 210	
17th 2	19th	3
Fundamentals: Pressure and Temperature	Fundamentals: Composition	
Key Topics	Key Topics	
1. Atmospheric pressure and density	1. Measures of abundance: mixing ratio,	
2. Equation of state (Ideal Gas Law)	number density, and partial pressure	
3. Changes of pressure with altitude	2. Composition of the present-day	
(Barometric Law)	atmosphere	
4. Temperature structure	Reading	
Reading	• <i>Jacob</i> (2023) §1	
• <i>Jacob</i> (2023) §2	• <i>Jacobson</i> (2012) \$1.3, 3.5-3.6	
• <i>Jacobson</i> (2012) §3.1-3.4	• Seinfeld and Pandis (2016) \$1.6-1.8	
 Seinfeld and Pandis (2016) \$1.1-1.5 	Note: Held Th Jan 19 @ 11:05 over Zoom	
Note: Held W Jan 18 @ 11:50 over Zoom		

TUESDAY	THURSDAY	
24th 4	26th 5	
Fundamentals: Mathematical Models	Fundamentals: Chemical Kinetics	
Key Topics	Key Topics	
1. Simple box models, budgets, and lifetime	1. Photodissociation	
2. Steady-state assumptions	2. Bi- and termolecular reaction rates	
3. Solution to the first-order loss problem	3. Equilibrium	
4. Continuity equation	4. Chemical families	
5. Multi-box models	5. Isotope fractionation	
Reading	Reading	
• <i>Jacob</i> (2023) §3	• <i>Jacob</i> (1999) §9	
	• <i>Jacobson</i> (2012) §1.4-1.5	
	• Seinfeld and Pandis (2016) \$3.1-3.6, 4.5	
	• Burkholder et al. (2015) §1.1, 2.1-2.6,	
	3.1-3.2, 4.1-4.2	
	PS1 Out	
31st 6	Feb 2nd7	
Biogeochemical Cycles: Oxygen and Nitrogen	Biogeochemical Cycles: Carbon	
Key Topics	Key Topics	
1. Global oxygen budget	1. Global carbon budget	
2. Global nitrogen budget	2. Carbon uptake by ocean	
Reading	3. Carbonate chemistry	
• <i>Jacob</i> (1999) \$6.1-6.4	Reading	
• <i>Jacobson</i> (2012) §2.3	• <i>Jacob</i> (1999) §6.5	
• <i>Kerr</i> (2005)	• Seinfeld and Pandis (2016) §22.2	
• <i>Kump</i> (2008)	PS1 Due at 7 PM; PS2 Out	
• Seinfeld and Pandis (2016) §2.3		
7th 8	9th 9	
Stratospheric Chemistry: Chapman	Stratospheric Chemistry: Catalytic Loss Cycles	
Mechanism	Key Topics	
Key Topics	1. HO_x cycles	
1. Stratospheric ozone formation	2. NO_x cycles	
Reading	3. Halogen cycles	
• <i>Jacob</i> (1999) §10.1	4. Stratospheric ozone depletion	
• Jacobson (2012) §11.1-11.3.1	Reading	
• Seinfeld and Pandis (2016) §5.1	• <i>Jacob</i> (1999) \$10.2	
• <i>Chapman</i> (1929)	• <i>Jacobson</i> (2012) §11.3.2-11.6	
	• Seinfeld and Pandis (2016) §5.2, 5.3, 5.4.1	
	• <i>WMO</i> (2014)	
	PS2 Due at 7 PM; PS3 Out	

TUESDAY	THURSDAY	
14th 10	16th 11	
 Stratospheric Chemistry: Heterogeneous Processes Key Topics Polar stratospheric clouds (PSCs) Antarctic ozone hole Volcanic aerosols and geoengineering Reading Jacob (1999) \$10.3-10.4 Jacobson (2012) \$11.7-11.9 Seinfeld and Pandis (2016) \$5.6-5.8 	Tropospheric Chemistry: BackgroundPhotochemistryKey Topics1. Photostationary state (NO, NO2, O3)2. OH production and budget3. HOx and NOx in the troposphere4. CO oxidation and production of ozone5. Tropospheric ozone budgetReading• Jacob (1999) \$11.1, 11.3-11.3.2, 11.4-11.5• Jacobson (2012) \$4.1-4.2.1• Seinfeld and Pandis (2016) \$6.1-6.3, 6.6PS3 Due at 7 PM; PS4 Out	
21st12 Tropospheric Chemistry: Oxidized Nitrogen Key Topics1. NO_x budget: sources, sinks, and trends2. Daytime NO_x chemistry3. Nighttime NO_x chemistry4. Reservoir species (NO_y) Reading• Jacob (1999) \$11.4• Jacobson (2012) \$4.2.2-4.2.3, 4.2.8• Seinfeld and Pandis (2016) \$6.5, 6.10-6.11	 23rd 13 Tropospheric Chemistry: Reduced Carbon Key Topics CO budget: sources, sinks, and trends CH₄ budget: sources, sinks, and trends Non-methane VOCs (isoprene, monoterpenes) Reading Jacob (1999) \$11.2, 11.3.3 Jacobson (2012) \$4.2.4-4.2.7 Seinfeld and Pandis (2016) \$6.4, 6.10-6.11 PS4 Due at 7 PM 	
28th 14	Mar 2nd 15	
 Tropospheric Chemistry: Aerosol Particles I 1. Particle composition, size distributions and morphology 2. Physiochemical properties of aerosol particles Reading Zhang et al. (2022) §1-2.2, 3 Jacobson (2012) §5.1-5.5 	 Tropospheric Chemistry: Aerosol Particles II 1. Growth and partitioning of organic aerosol particles 2. Multiphase and heterogeneous reactions Reading Zhang et al. (2022) §4-5 Jacobson (2012) §5.1-5.5 PS5 Due at 7 PM; PS6 Out 	
7th Spring Break	9th Spring Break	

TUESDAY	THURSDAY
14th 16	16th 17
Tropospheric Chemistry: Halogens Chlorine Bromine Iodine Reading Wang et al. (2021) 	Air Pollution: Particulate Matter Key Topics 1. Human health effects of aerosol particles 2. Aerosol particles and urban air pollution 3. Acid rain impacts, chemistry, and physics 4. Atmospheric fertilization: eutrophication and stresses Reading • Zhang et al. (2022) §6.3-6.7 • Jacob (1999) §13 • Gruber and Galloway (2008) • Jacobson (2012) §10 • Seinfeld and Pandis (2016) §20.5-20.6 • Bowman et al. (2008)
 Air Pollution: Photochemical Smog Key Topics Surface ozone pollution: impacts Urban O₃ pollution chemistry NO_x- vs. VOC-limited regimes Reading Jacob (1999) \$12 Jacobson (2012) \$4.3-4.5 Seinfeld and Pandis (2016) \$6.4, 6.10-6.11 Sillman (2002) National Research Council Committee on Tropospheric Ozone (1991) 	Midterm In Class PS7 Out
 28th 19 Air Pollution: Heavy Metals and Persistents Key Topics Lead: Impacts and global budget Mercury: Impacts, budget and chemistry Persistent organic pollutants (POPs) Perfluoroalkyl and polyfluoroalkyl substances (PFAS) Readings Riva et al. (2012) Selin (2009) UNEP/AMAP (2011) §1-3, 7 Chambers et al. (2021) Gaffney and Marley (2014) 	30th20Air Pollution: Impact of MeteorologyKey Topics1. Forces, winds and global circulation2. Vertical pollutant transport3. Horizontal pollutant transportReading• Jacob (2023) §4• Jacobson (2012) §6• Seinfeld and Pandis (2016) §16, 21PS7 Due at 7 PM; PS8 Out

TUESDAY	THURSDAY	
Apr 4th 21	6th 22	
Air Pollution: Indoor Air Quality	Atmospheric Radiation: Shortwave	
1. Photochemical pollutants	Key Topics	
2. Carbon monoxide	1. Solar radiation spectrum	
3. Radon	2. Atmospheric scattering of radiation	
4. Asbestos	3. Visibility	
Reading	4. Colors in the atmosphere	
• <i>Tran et al.</i> (2020)	Reading	
• <i>Tham</i> (2016)	• <i>Jacob</i> (1999) §8.2	
• <i>Jacobson</i> (2012) §9	• <i>Jacobson</i> (2012) §7	
	PS8 Due at 7 PM; PS9 Out	
11th 23	13th 24	
Atmospheric Radiation: Longwave	Chemistry-Climate Interactions: Chemical	
Key Topics	Forcing of Climate	
1. Atmospheric absorption of radiation	Key Topics	
2. Black body radiation (Planck's Law)	1. Radiative forcing	
3. Earth's radiative balance and effective	2. Global warming potential (GWP)	
temperature	3. Global temperature potential (GTP)	
4. The greenhouse effect	4. Climate feedbacks	
Reading	5. Forcing agents	
• <i>Jacob</i> (1999) \$7.1-7.3	Reading	
 Jacobson (2012) \$12.1-12.2 	• <i>IPCC</i> (2021) \$6.1-6.4, \$7 Exec. Summ.	
	• <i>Jacob</i> (1999) §7.4-7.6	
	• Zhang et al. (2022) §6.2	
	• <i>Jacobson</i> (2012) §12.3-12.7	
	PS9 Due at 7 PM; PS10 Out	
18th 25	20th 26	
Chemistry-Climate Interactions: Climate	Air Quality Policy: Regulations	
Forcing of Chemistry	Key Topics	
Key Topics	1. Regulation in the United States (the Clean	
1. Anticipated future emissions	Air Act and its Amendments)	
2. Changing chemical regimes and	2. Regulation and pollution trends	
air-quality "climate penalties"	elsewhere in the world	
3. Mitigation strategies	3. International treaties	
4. The future of air pollution	Reading	
Reading	• <i>Kuklinska et al.</i> (2015)	
• <i>IPCC</i> (2021) \$4 Exec. Summ., \$6.5-6.8	• <i>OECD</i> (2017) (Air Pollution Sections)	
• Jacob and Winner (2009)	• <i>Jacobson</i> (2012) §8	
• <i>Shindell et al.</i> (2012)		
Seinfeld and Pandis (2016) §23.10		
• <i>Fiore et al.</i> (2012, 2015)		

TUESDAY	THURSDAY
25th 27	27th
Air Quality Policy: Technology	Reading Period
Key Topics	
1. Atmospheric measurement techniques	
2. Pollution control technologies	
3. Clean, low-risk, sustainable energy	
systems	
Reading	
 MJ Bradley & Associates (2005) 	
• <i>Heard</i> (2006)	
• <i>Jacobson</i> (2012) §13	
PS10 Due at 7 PM	
May 2nd 28	4th 29
Final Exam Date TBD by Registrar	

2.1 RECITATION

In addition to lectures, there will be a 1-hr recitation session held each week, in which the TA will go over example problems similar to the homework and exams, and answer any questions from the class.

3 READINGS

3.1 REQUIRED

Воок

- IPCC (2021), *Climate Change 2021: The Physical Science Basis. Contribution of Working Group I to the Sixth Assessment Report of the Intergovernmental Panel on Climate Change*, 2391 pp., Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, doi:10.1017/9781009157896.
- Jacob, D. J. (1999), Introduction to Atmospheric Chemistry, 1st ed., Princeton University Press, Princeton, NJ, https://acmg.seas.harvard.edu/education/ introduction-atmospheric-chemistry.
- Jacob, D. J. (2023), Introduction to Atmospheric Chemistry, 2nd ed., https://acmg.seas.harvard.edu/education/2nd-edition.
- Zhang, Y., P. Liu, Y. Han, Y. Li, Q. Chen, M. Kuwata, and S. T. Martin (2022), *Aerosols in Atmospheric Chemistry*, American Chemical Society, Washington, DC, USA, doi:10.1021/acsinfocus.7e5020.

ARTICLES AND EXCERPTS

Chambers, W., J. Hopkins, and S. Richards (2021), A review of per- and polyfluorinated alkyl substance impairment of reproduction., *Front Toxicol*, *3*, 732,436, doi:10.3389/ftox.2021.732436.

- Gruber, N., and J. N. Galloway (2008), An Earth-system perspective of the global nitrogen cycle, *Nature*, *451*(7176), 293–296, doi:10.1038/nature06592.
- Jacob, D. J., and D. A. Winner (2009), Effect of climate change on air quality, *Atmos Environ*, 43(1), 51–63, doi:10.1016/j.atmosenv.2008.09.051.
- Kuklinska, K., L. Wolska, and J. Namiesnik (2015), Air quality policy in the U.S. and the EU a review, *Atmos Poll Res*, 6(1), 129–137, doi:10.5094/APR.2015.015.
- MJ Bradley & Associates (2005), Best Available Technology for Air Pollution Control: Analysis Guidance and Case Studies for North America, *Tech. rep.*, Manchester, NH, http://www3.cec.org/ islandora/en/item/2195-best-available-technology-air-pollution-control/.
- Riva, M., A. Lafranconi, M. D'Orso, and G. Cesana (2012), Lead poisoning: historical aspects of a paradigmatic occupational and environmental disease., *Saf Health Work*, *3*(1), 11–16, doi:10.5491/SHAW.2012.3.1.11.
- Selin, N. E. (2009), Global Biogeochemical Cycling of Mercury: A Review, *Annu Rev Env Resour*, *34*(1), 43–63, doi:10.1146/annurev.environ.051308.084314.
- Tran, V., D. Park, and Y. Lee (2020), Indoor air pollution, related human diseases, and recent trends in the control and improvement of indoor air quality., *Int J Environ Res Public Health*, *17*(8), 2927, doi:10.3390/ijerph17082927.
- UNEP/AMAP (2011), *Climate Change and POPs: Predicting the Impacts*, 62 pp., Report of the UNEP/AMAP Expert Group. Secretariat of the Stockholm Convention, Geneva, https://www.amap.no/documents/doc/climate-change-and-pops-predicting-the-impacts/753.
- Wang, X., et al. (2021), Global tropospheric halogen (Cl, Br, I) chemistry and its impact on oxidants, *Atmospheric Chemistry and Physics*, *21*(18), 13,973–13,996, doi:10.5194/acp-21-13973-2021.
- Zahnle, K., L. Schaefer, and B. Fegley (2010), Earths earliest atmospheres., *Cold Spring Harb Perspect Biol*, *2*(10), a004,895, doi:10.1101/cshperspect.a004895.
- Zhang, Y., P. Liu, Y. Han, Y. Li, Q. Chen, M. Kuwata, and S. T. Martin (2022), *Aerosols in Atmospheric Chemistry*, American Chemical Society, Washington, DC, USA, doi:10.1021/acsinfocus.7e5020.

3.2 SUPPLEMENTAL

BOOKS

- Heard, D. E. (Ed.) (2006), *Analytical Techniques for Atmospheric Measurement*, Blackwell Publishing, Oxford, UK, doi:10.1002/9780470988510.
- Jacobson, M. Z. (2012), *Air Pollution and Global Warming: History, Science, and Solutions*, 2 ed., Cambridge University Press, Cambridge, doi:10.1017/CBO9781139109444.
- OECD (2017), Best Available Techniques (BAT) to Prevent and Control Industrial Pollution: Policies on BAT or similar concepts Across the World, vol. OECD Environment, Health and Safety Publications Series on Risk Management, Organisation for Economic Cooperation and Development, Paris, https://www.oecd.org/chemicalsafety/risk-management/ policies-on-best-available-techniques-or-similar-concepts-around-the-world.pdf.

- Seinfeld, J. H., and S. N. Pandis (2016), Atmospheric Chemistry and Physics: From Air Pollution to Climate Change, 1152 pp., Wiley, Hoboken, NJ, https://ebookcentral.proquest.com/lib/ rochester/detail.action?docID=4462549.
- Zhang, Y., P. Liu, Y. Han, Y. Li, Q. Chen, M. Kuwata, and S. T. Martin (2022), *Aerosols in Atmospheric Chemistry*, American Chemical Society, Washington, DC, USA, doi:10.1021/acsinfocus.7e5020.

ARTICLES

- Bowman, W. D., C. C. Cleveland, L. Halada, J. Hreško, and J. S. Baron (2008), Negative impact of nitrogen deposition on soil buffering capacity, *Nat Geosci*, *1*(11), 767–770, doi:10.1038/ngeo339.
- Burkholder, J. B., et al. (2015), Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies: Evaluation Number 18, *Tech. Rep. JPL Publication 15-10*, Jet Propulsion Laboratory, Pasadena, CA, https://jpldataeval.jpl.nasa.gov/pdf/JPL_Publication_15-10.pdf.
- Chapman, S. (1929), A theory of upper-atmospheric ozone, *Memoirs of the Royal Meteorological Society*, 3(26), 103–125.
- Fiore, A. M., et al. (2012), Global air quality and climate, *Chem Soc Rev*, 41(19), 6663–6683, doi:10.1039/c2cs35095e.
- Fiore, A. M., V. Naik, and E. M. Leibensperger (2015), Air Quality and Climate Connections, *JAPCA J Air Waste Ma*, 65(6), 645–685, doi:10.1080/10962247.2015.1040526.
- Gaffney, J., and N. Marley (2014), In-depth review of atmospheric mercury: sources, transformations, and potential sinks, *Energy and Emission Control Technologies*, p. 1, doi:10.2147/eect.s37038.
- Kerr, R. A. (2005), The Story of O₂, *Science*, 308(5729), 1730–1732, doi:10.1126/science.308.5729.1730.
- Kump, L. R. (2008), The rise of atmospheric oxygen, *Nature*, 451(7176), 277–278, doi:10.1038/nature06587.
- National Research Council Committee on Tropospheric Ozone (1991), *Rethinking the Ozone Problem in Urban and Regional Air Pollution*, National Academy Press, Washington, DC, doi:10.17226/1889.
- Seinfeld, J. H., and S. N. Pandis (2016), Atmospheric Chemistry and Physics: From Air Pollution to Climate Change, 1152 pp., Wiley, Hoboken, NJ, https://ebookcentral.proquest.com/lib/rochester/detail.action?docID=4462549.
- Shindell, D. T., et al. (2012), Simultaneously mitigating near-term climate change and improving human health and food security., *Science*, *335*(6065), 183–189, doi:10.1126/science.1210026.
- Sillman, S. (2002), Chapter 12 The relation between ozone, NO_x and hydrocarbons in urban and polluted rural environments, in *Air Pollution Science for the 21st Century*, pp. 339–385, Elsevier, doi:10.1016/S1474-8177(02)80015-8.
- Tham, K. W. (2016), Indoor air quality and its effects on humans—A review of challenges and developments in the last 30 years, *Energy and Buildings*, *130*, 637–650, doi:10.1016/j.enbuild.2016.08.071.
- WMO (2014), Scientific Assessment of Ozone Depletion: 2014, *Tech. Rep. Report No. 55*, Geneva, Switzerland.

4 GRADING

Your final grade will be calculated with the following breakdown

Problem Sets:	40~%
Midterm:	20 %
Final Exam:	25~%
Quizzes:	10~%
Participation:	5 %
Total:	100 %

4.1 PROBLEM SETS

The aim of the problem sets is to help you learn the course concepts. **Working together with your classmates is thus strongly encouraged**, although problem sets should always be solved and written up individually. **If you collaborate, write who you worked with on your submission**.

Show all work, explaining in sufficient detail how you arrived at the answer. Describe the rational behind each step using language like "Convert from kg to molecules" or "Apply the hypsometric equation to determine layer thickness." Partial credit for ultimately wrong answers will be assigned based on work shown. A correct answer with no work shown earns no credit. A numerical answer without units is incorrect.

Problem sets are due to Gradescope at 7 PM on the day indicated. After that, 10% is deducted off the possible total score for each day late. No credit is given after one week late.

Your lowest problem set score will be dropped from your final grade calculation. You may use this at your discretion as a free pass for a busy week or to try to maximize your mean problem set grade.

No exceptions to the late policy will be accepted, except with a medical note or request via the disabilities office.

4.2 EXAMS

All exams will be comprised of a multiple-choice section, a short-answer section, and a longer quantitative problem section.

4.2.1 MIDTERM

There will be one midterm exam administered in class, which will test qualitative content from Lectures 1-18 and quantitative questions derived from Problem Sets 1-6.

4.2.2 FINAL

There will be one comprehensive final exam. The final will contain content from throughout the course, with a greater emphasis on the second half. Quantitative questions will be derived from the Problem Sets.

4.3 QUIZZES

At the end of each class, there will be a short quiz on the main concepts covered by the reading and/or lecture. Please bring calculators to class to use for in-class problems. Unexcused absences will result in a zero on that day's quiz. Quizzes on days of an excused absence (see below) will have zero weight in final quiz grade.

4.4 PARTICIPATION

The participation score will reflect the following activities:

- 1. Attendance to class as reflected by quiz participation, and attendance to recitation. I must be informed via e-mail of any absences ahead of time. Illness, other educational/research experiences, and observance of religious holidays are examples of excusable absences.
- 2. General participation: e.g., by asking or answering questions during class, and/or by asking questions during office hours.

5 CLASSROOM POLICIES

5.1 DIVERSITY AND INCLUSION

This class is an inclusive and welcoming learning environment for all students regardless of background or ability, consistent with University policy, state and federal laws and the instructors' personal beliefs. Students must respect the different experiences, identities, beliefs and values expressed by their peers, and refrain from derogatory comments about other individuals, cultures, groups, or viewpoints.

Please make sure that I am pronouncing your name correctly, and let me know if you have any preferred nicknames and/or pronouns that you would like me to use.

In the event you encounter any barrier(s) to full participation in this course due to the impact of a disability, please contact the Office of Disability Resources. The access coordinators in the Office of Disability Resources can meet with you to discuss the barriers you are experiencing and explain the eligibility process for establishing academic accommodations. You can reach the Office of Disability Resources at: disability@rochester.edu; (585) 276-5075; Taylor Hall; http://www.rochester.edu/ college/disability.

5.2 COVID-19

You are expected to adhere to all University policies regarding the COVID-19 pandemic (vaccination, masking, distancing, staying home if you have any symptoms, etc.). Please make sure you stay up to date on these policies, as they continually evolve.

Given the extraordinary circumstances of the ongoing COVID-19 pandemic, I acknowledge that students may be subject to a host of pressures and difficulties that will make learning this semester difficult. I encourage you to meet with me about any concern or situation that affects your ability to complete your academic work successfully.

5.3 Electronic Devices

Cell phones must be silenced. You may use a laptop or tablet to take notes. Mobile devices and laptops may also be used for prepared active learning activities. However, they should not be used for anything else during class. If a peer tells me your actions on your laptop are distracting during a lecture, you will lose the privilege.

6 Academic Honesty

All assignments and activities associated with this course must be performed in accordance with the University of Rochester's Academic Honesty Policy. A comprehensive description of the University of Rochester's Academic Honesty Policy is available at: http://www.rochester.edu/college/honesty. For this course, all exams and reports need to be completed individually, but I encourage collaboration on the problem sets.

7 INITIAL MEETING

I would like to learn about your background and goals, both for this course and for the future, to help tailor the semester to those interests. You are encouraged to schedule (via e-mail) a quick 5-minute meeting early in the semester. The meeting is entirely optional and will not impact your participation grade.

8 FEEDBACK

I want you to get the most out of this class. Students are encouraged to offer feedback at any time about the course and my instruction to me in person, through e-mail to <u>lee.murray@rochester.edu</u>, or via an anonymous note placed in my departmental mailbox located in Hutchison Hall 227. At the end of the course, I would greatly appreciate if you were to fill out the course review.